



PTYS544

Physics of the High Atmosphere

Basic details

👁 Location / Time

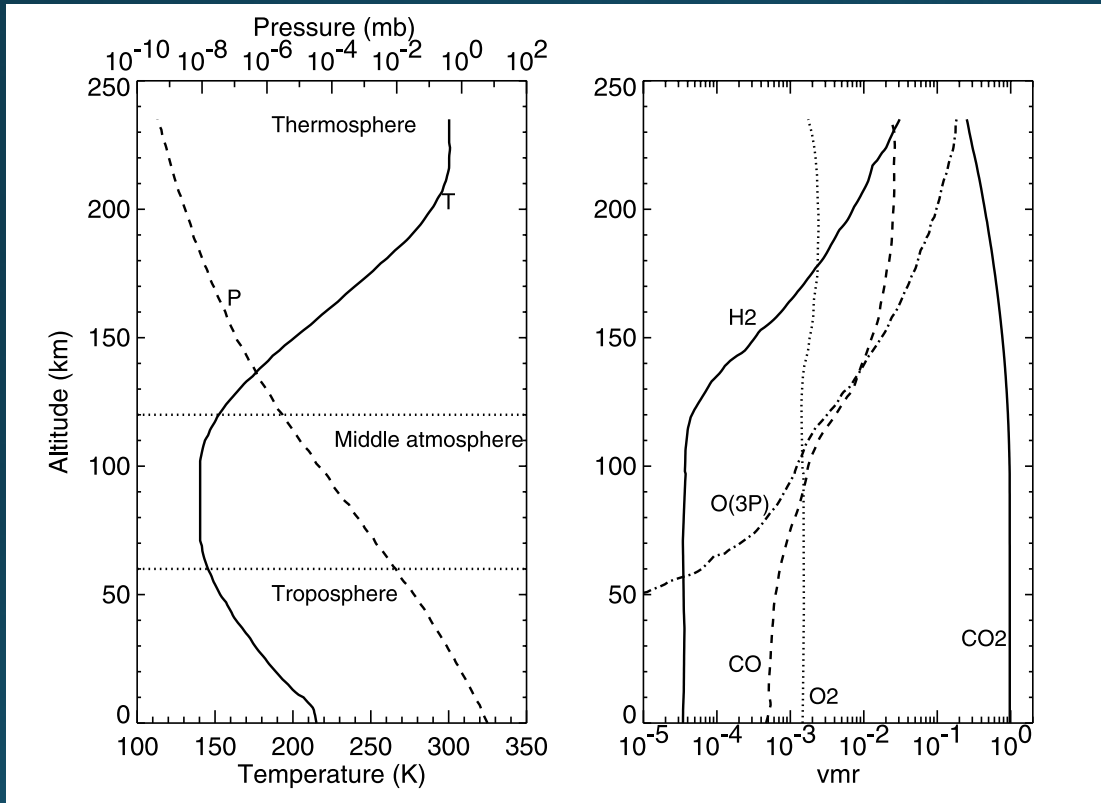
- Tuesday & Thursday, 12:30 – 13:45
- Kuiper Space Science (KSS)

👁 Instructor

- Tommi Koskinen, KSS 421
- tommik@email.arizona.edu



Mars: Basic atmospheric structure



- Middle atmosphere energy balance:
- Absorption in the CO₂ near-IR bands
 - Absorption and emission in the CO₂ 15- μ m band
 - Absorption and scattering by dust

Plot from Gonzales-Galindo et al. (2005)

Temperatures from MAVEN/NGIMS

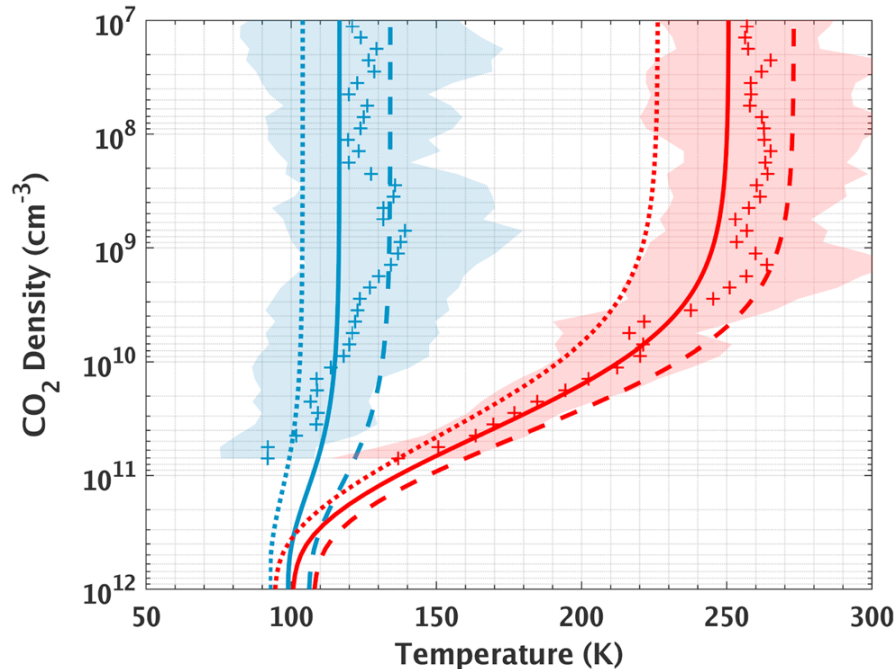


Figure 19. Comparison of modeled temperature profiles with NGIMS data. The model temperature profiles for equatorial latitudes at 12 p.m. (red lines) and 12 a.m. (blue lines) compared to DD2 (red +) and DD6 (blue +) NGIMS temperature profiles. The solid lines are for models calculated with the nominal O-CO₂ collisional deexcitation rate of $k = 3 \times 10^{-12} \text{ cm}^{-3} \text{ s}^{-1}$, the dotted lines for $k = 6 \times 10^{-12} \text{ cm}^{-3} \text{ s}^{-1}$, and the dashed lines for $k = 1.5 \times 10^{-12} \text{ cm}^{-3} \text{ s}^{-1}$. The shaded regions represent 1σ variabilities of the NGIMS temperatures. NGIMS = Neutral Gas and Ion Mass Spectrometer; DD = Deep Dip.

From Stone et al.
(2018)

Note that the expected solar cycle and seasonal variation in dayside exospheric temperature is about 150 K (Bougher et al. 2015a).

Temperature peak and time lag

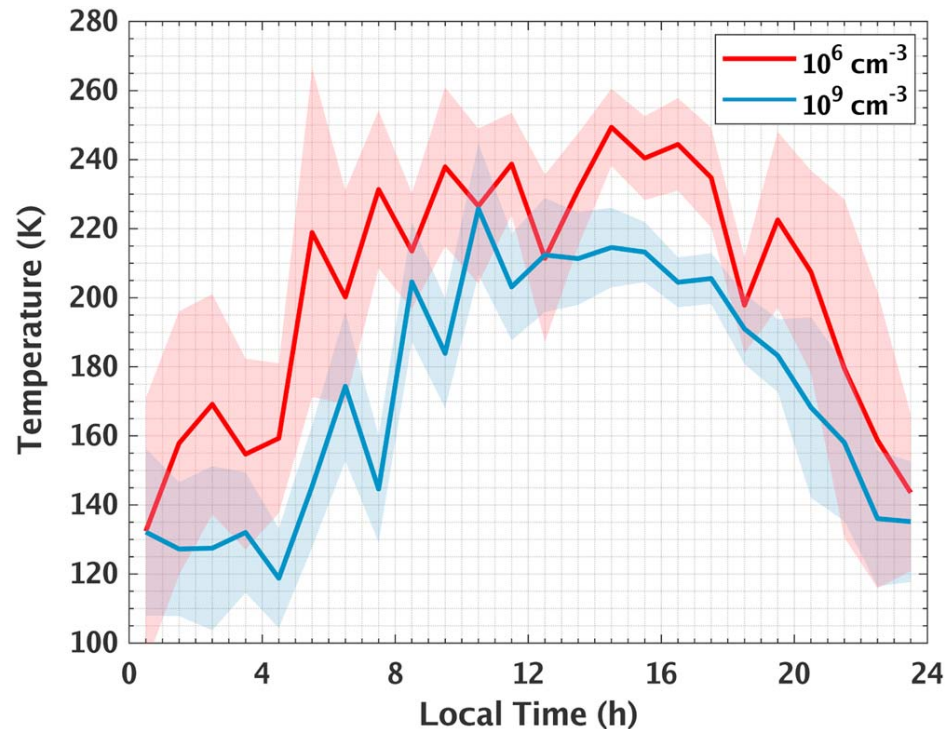
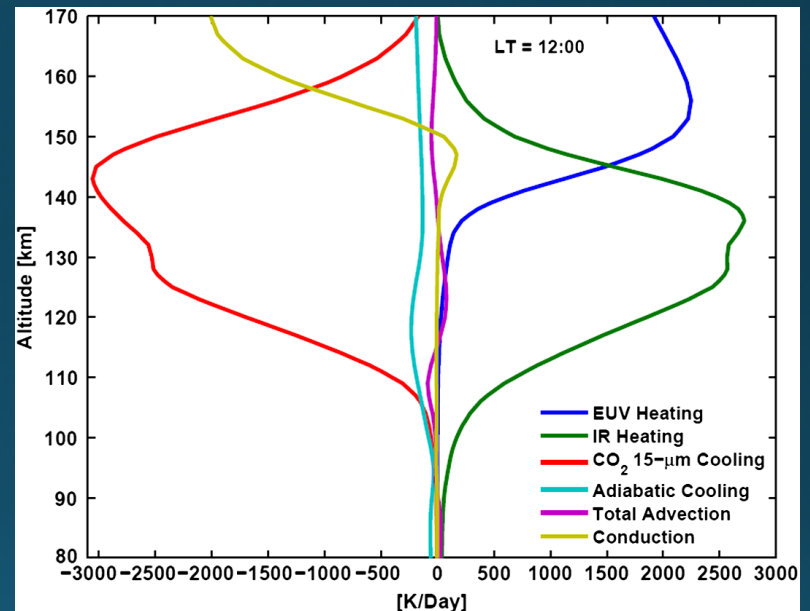
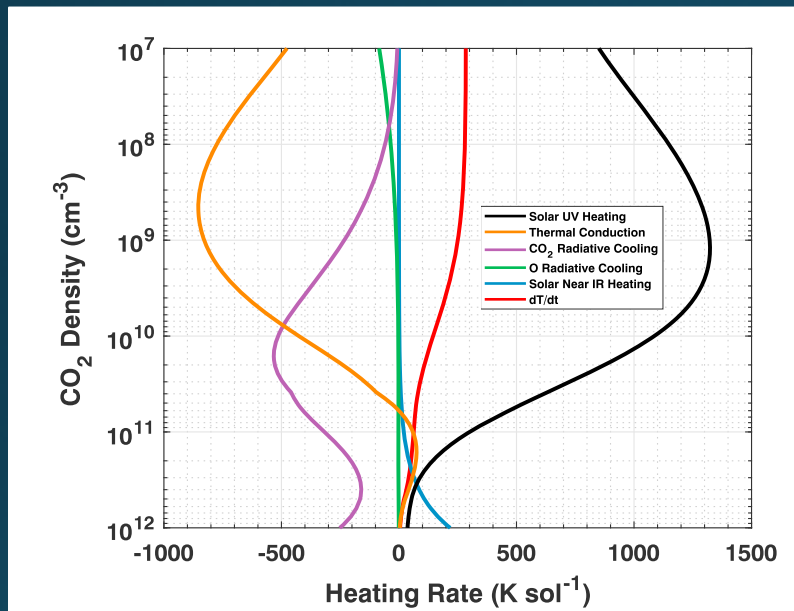


Figure 16. Temperature as a function of Martian local time at two constant CO₂ density levels: 10⁶ cm⁻³ (red) and 10⁹ cm⁻³ (blue). Measurements are constrained to latitudes between 60°N and 60°S. The shaded regions represent 1 σ variabilities.

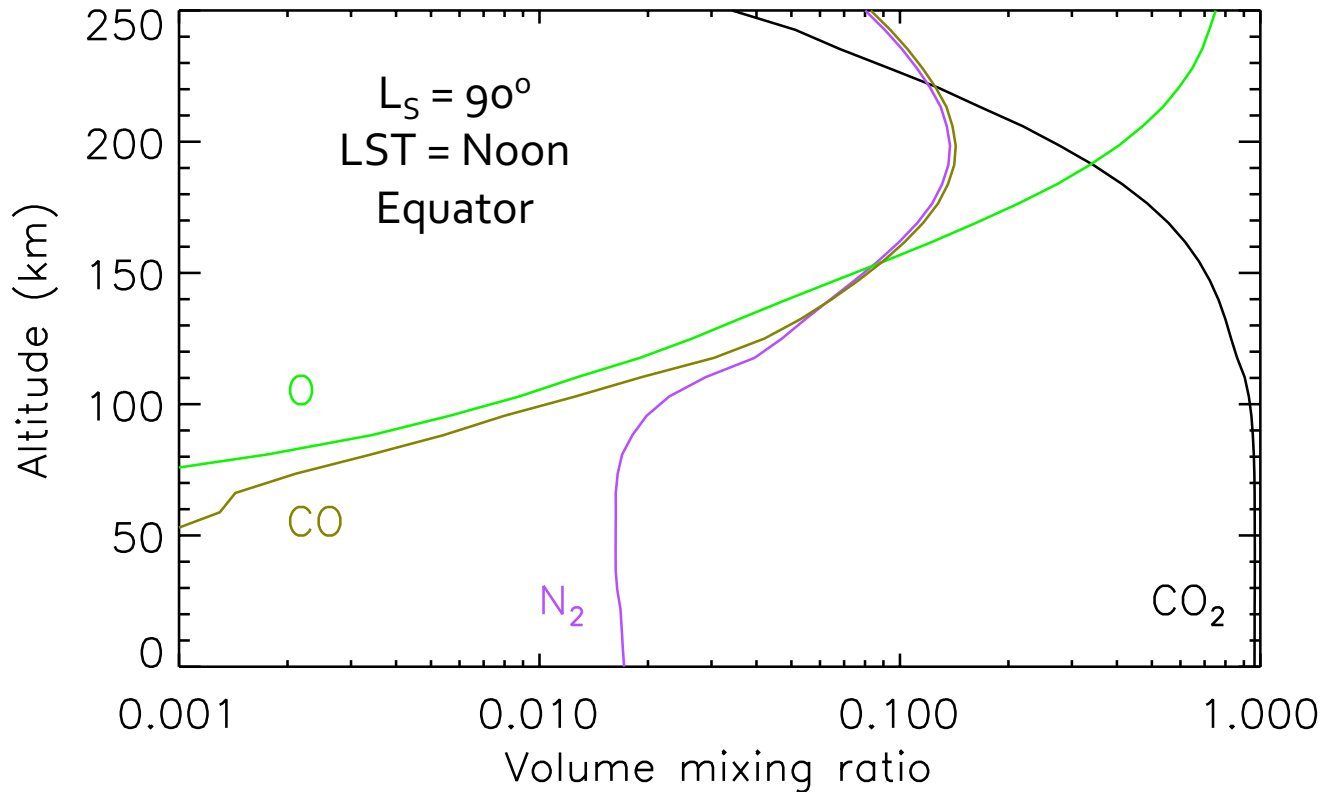
From Stone et al. (2018)

Compare with Venus



The CO₂ thermostat is less efficient on Mars, probably because the Martian O/CO₂ density ratio is smaller (Garcia Munoz et al. 2017).

Neutral composition



Results from the Mars Climate Database (MCD):
http://www-mars.lmd.jussieu.fr/mcd_python/

Neutral photochemistry

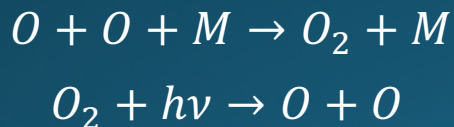
Photolysis of CO₂ in the stratosphere
(above ~60 km):



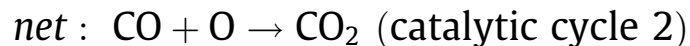
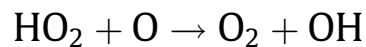
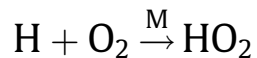
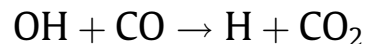
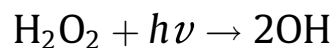
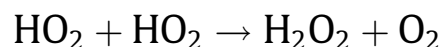
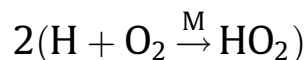
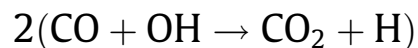
The reverse reaction is spin-
forbidden and thus extremely slow:



Instead, we might expect:

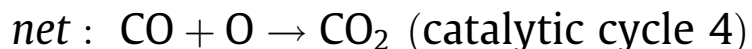
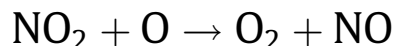
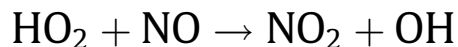
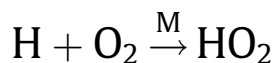
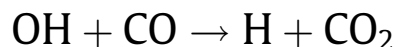
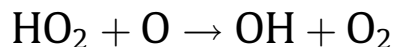
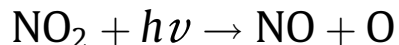
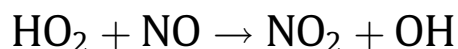
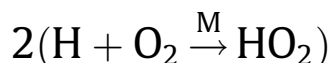
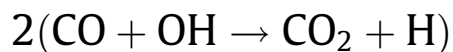


Should end up with CO, O₂ and O
with [CO]/[O₂] = 2.



McElroy and Donahue (1972);
Parkinson and Hunten (1972)

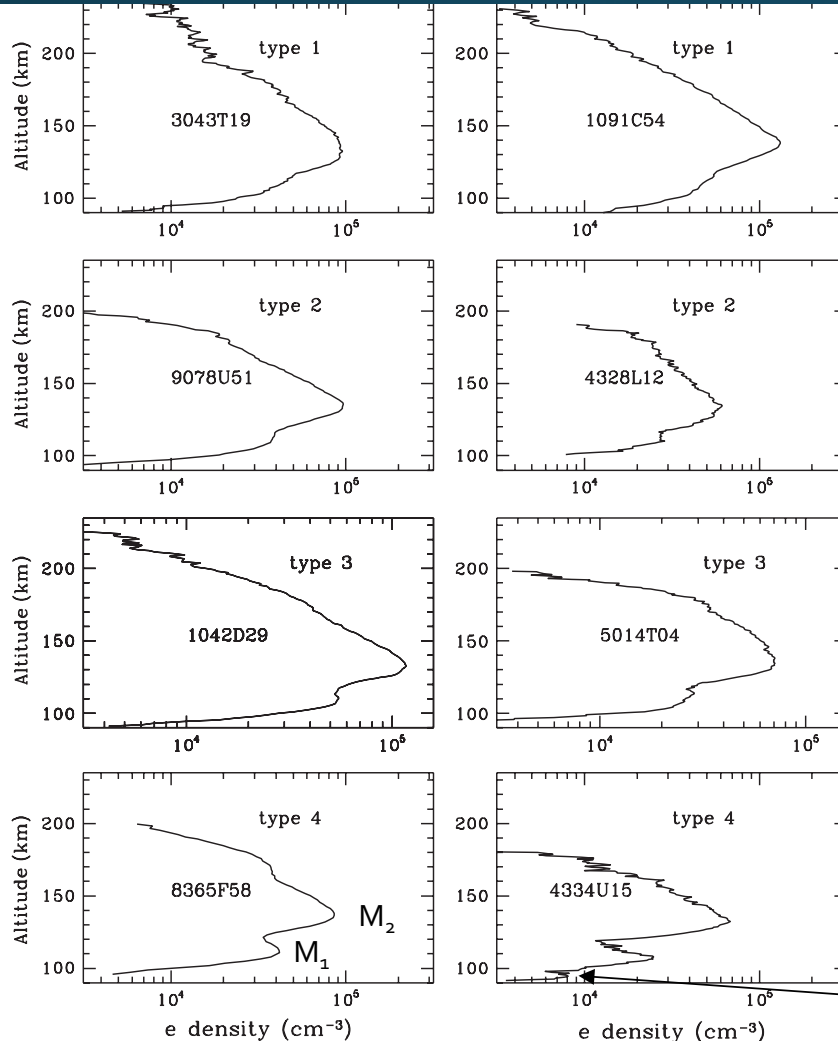
Neutral photochemistry



See also some new ideas about the possible role of HOCO radicals (Boxe et al. 2014)

Nair et al. (1994); Yung and Demore (1999)

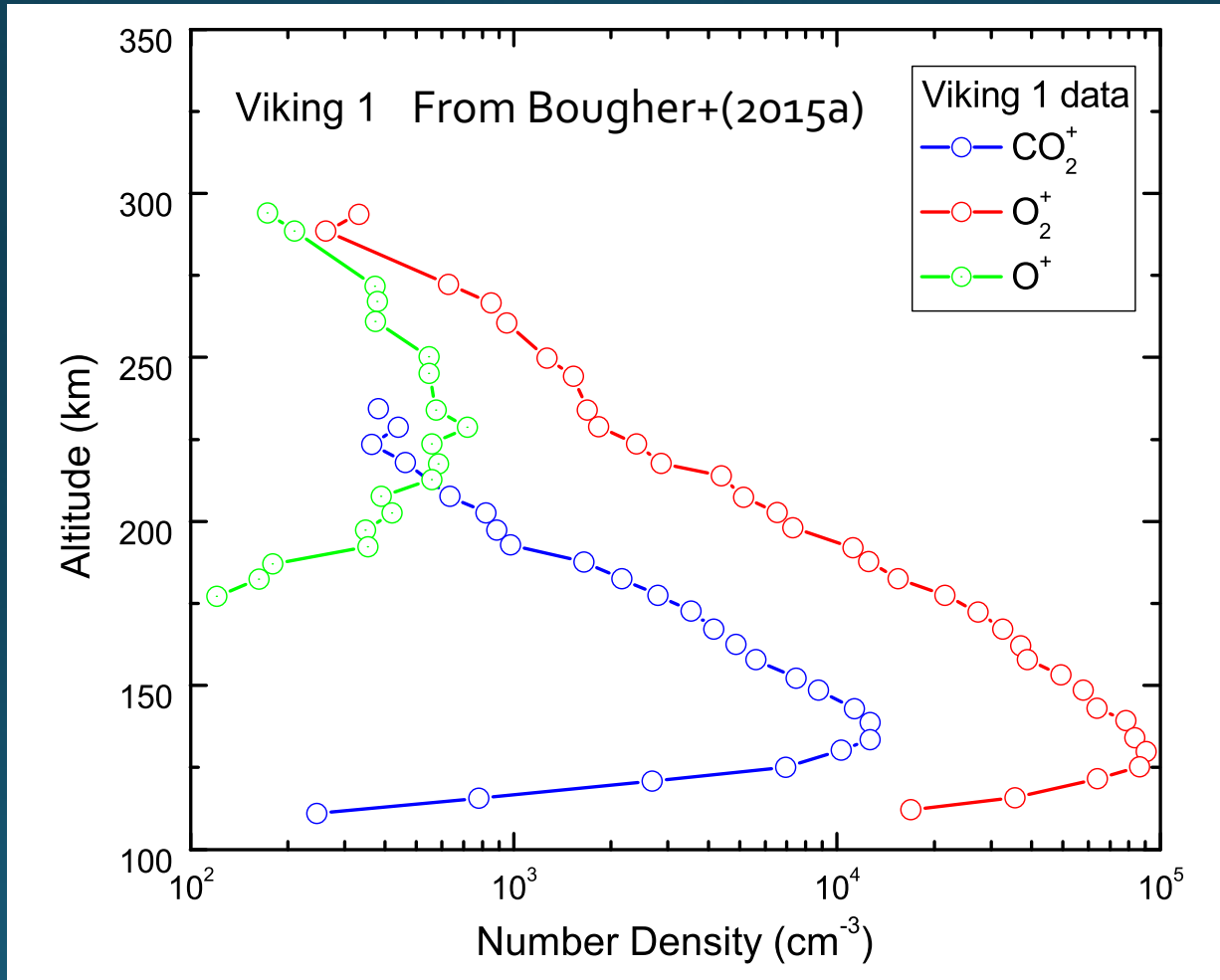
Electron densities



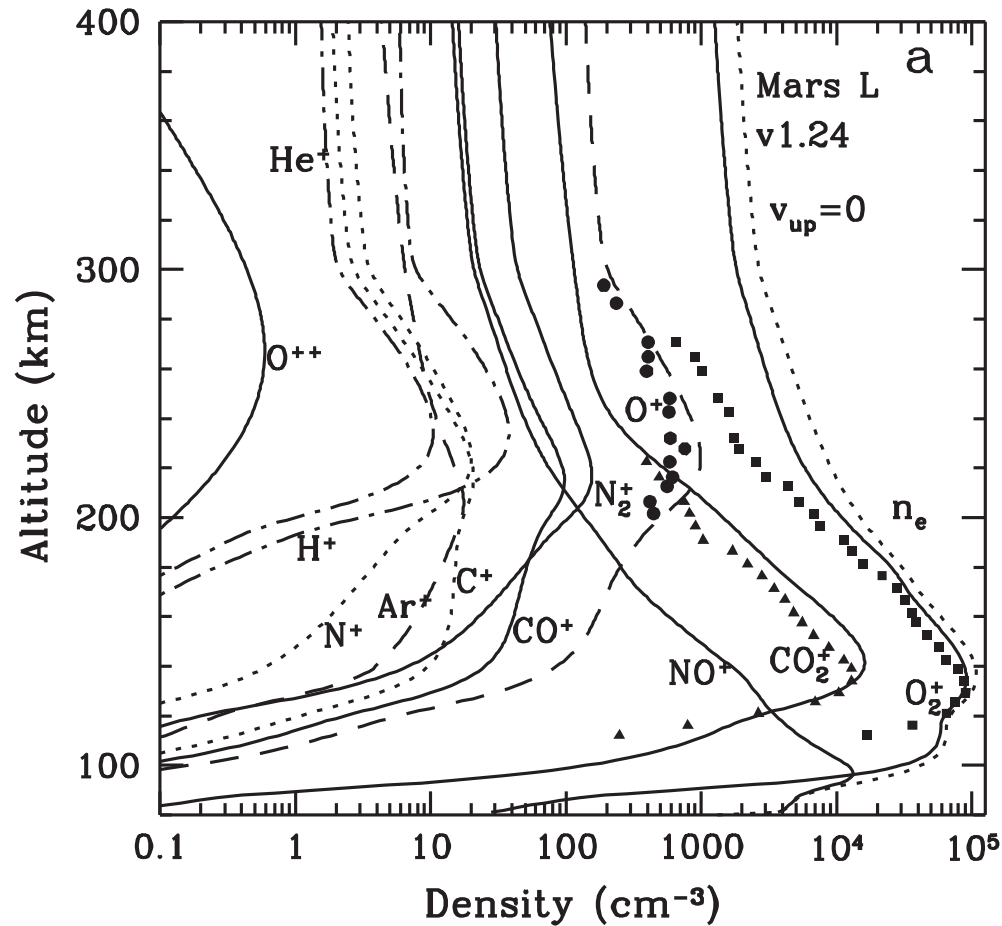
Main peak due to photoionization, lower peak from photoelectron impact ionization, third low altitude peak due to meteoritic ions (Mg^+ , Na^+ , Fe^+). Plot of Mars Global Surveyor electron density profiles from Fox and Weber (2012).

Meteoritic ions

Ionosphere composition

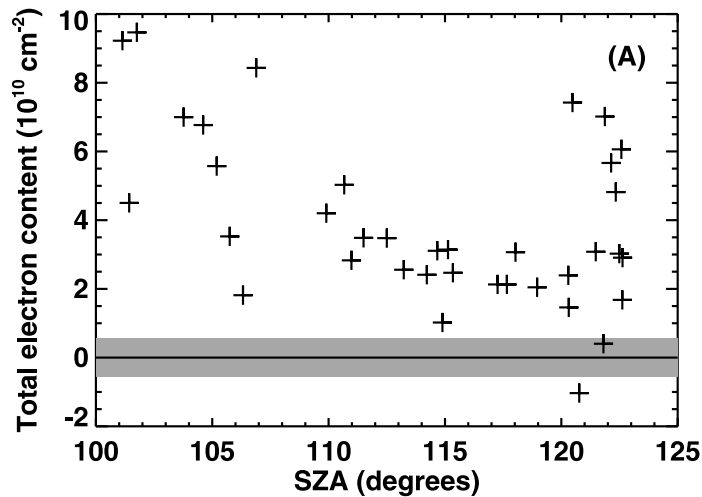
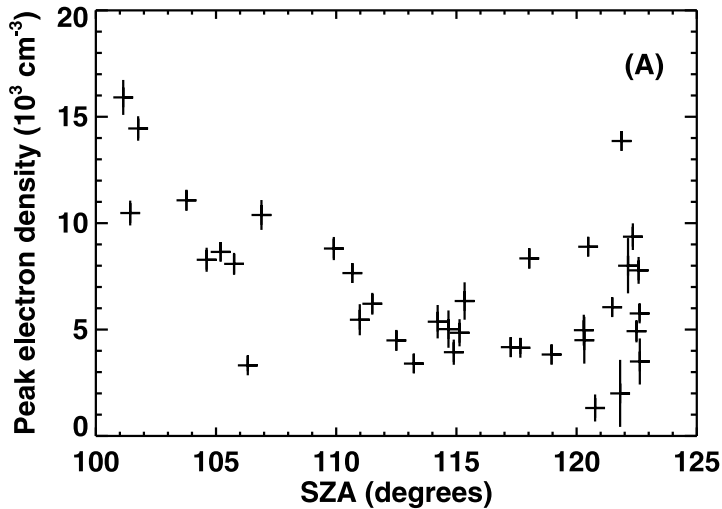


Model ionosphere



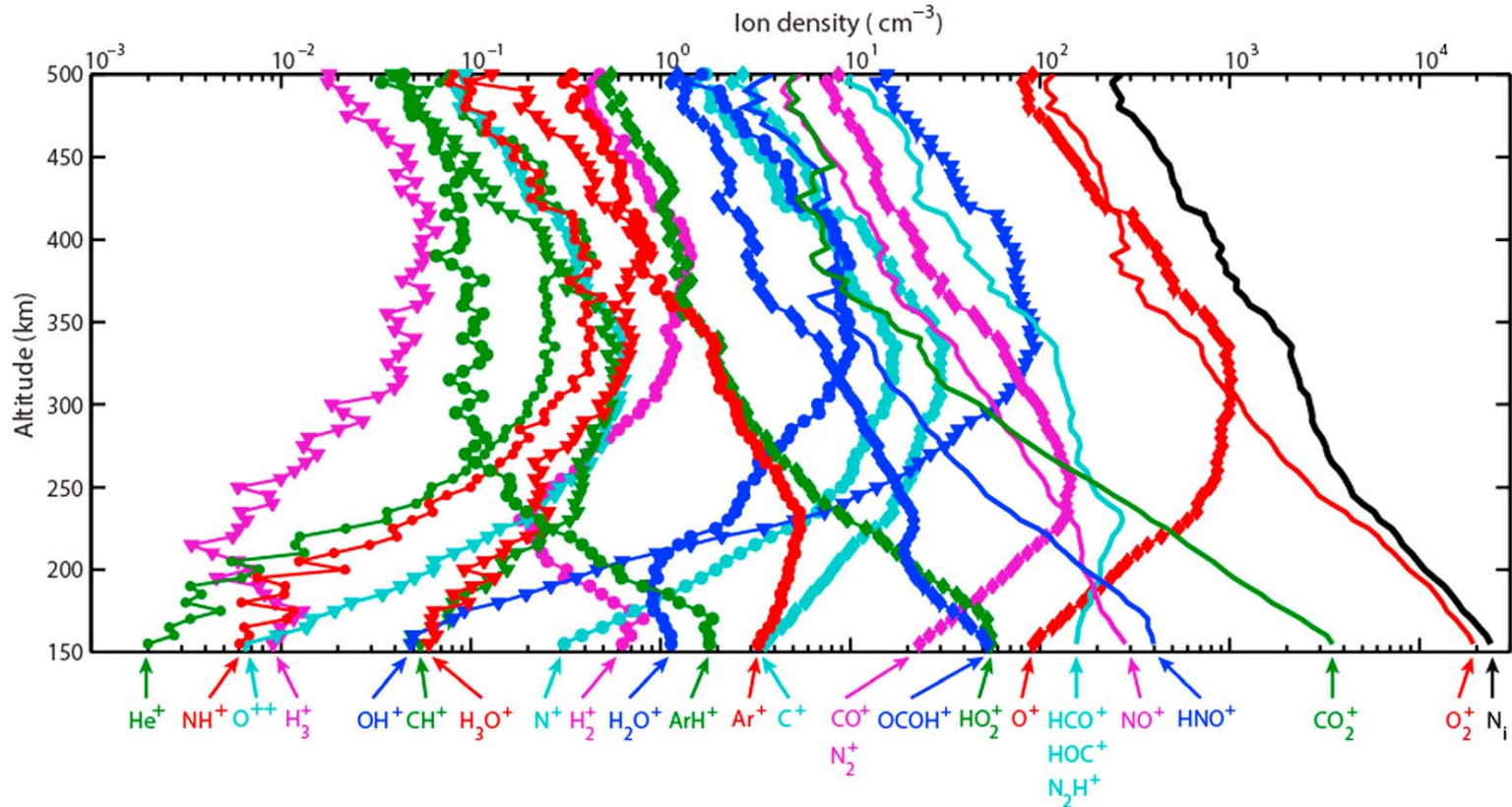
From Fox (2015).
Sources and sinks of
the main ions similar
to Venus.

Day/night differences



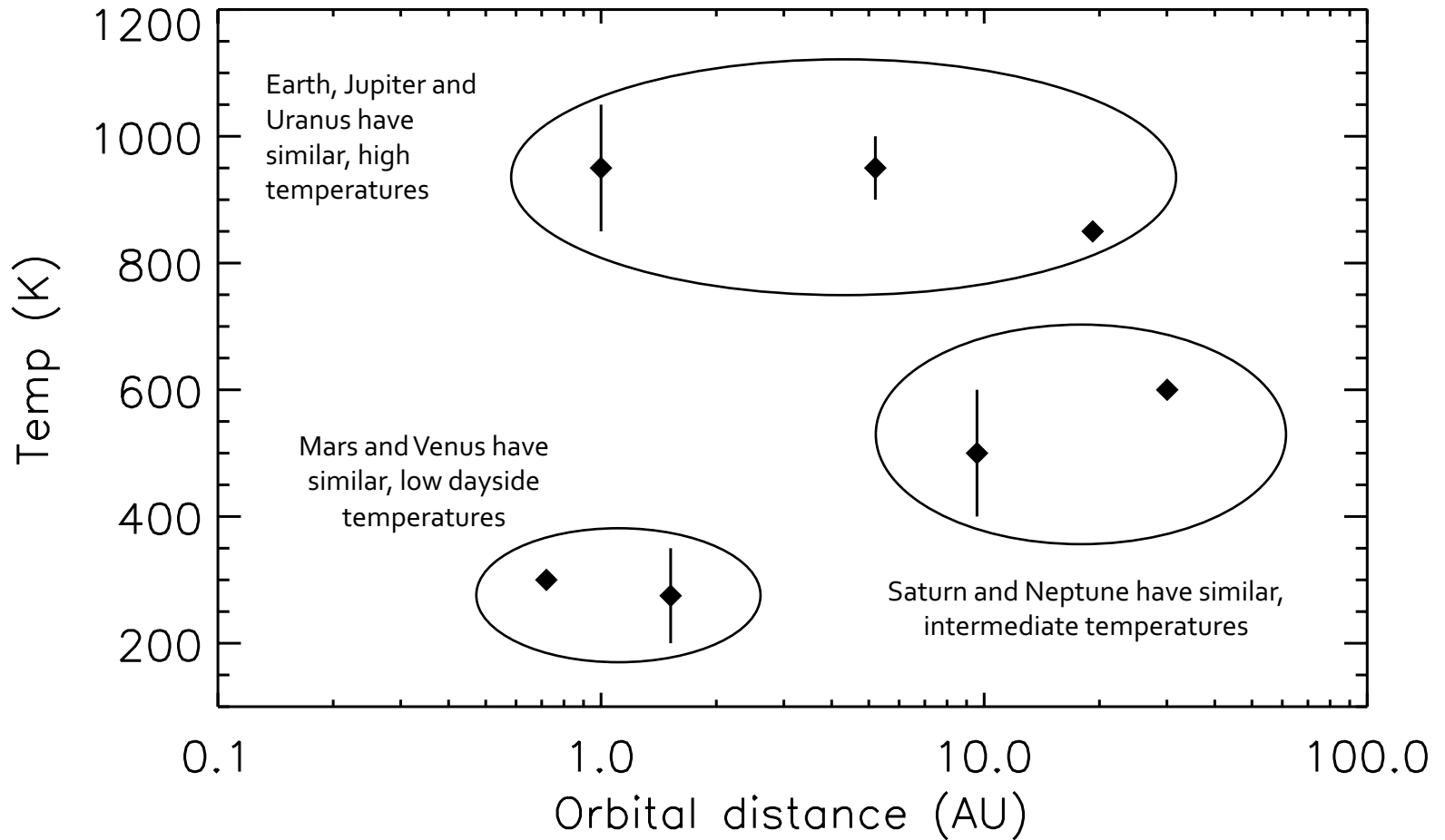
Night side plasma from day to night transport and electron precipitation. See study by Withers et al. (2012) based on Mars Express Radio Science results, and Benna et al. (2015) for SZA variations in ion densities.

Ionosphere from MAVEN/NGIMS



From Benna et al. (2015)

Exospheric temperatures



Auroral electrodynamics

Discrete aurora: dance of the spirits

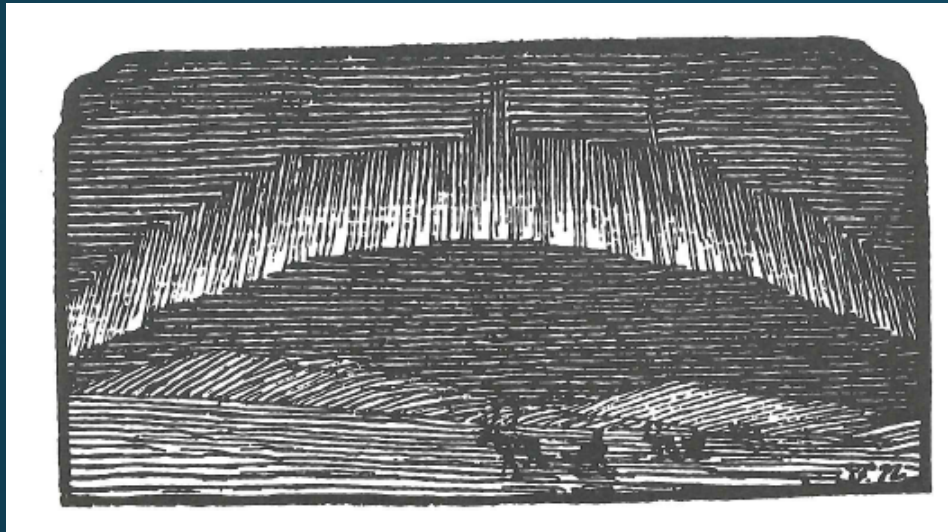


The death dance of the spirits,
warriors with their plumes and war
clubs, flaring away to the northward,
in the frosty nights of winter.

Hiawatha, H. W. Longfellow



The auroral arc



“Pohjola’s gates are in sight,
the evil gateways glitter,
the bright covers are
glowing...”

Finnish folk poetry, Sampo

Wood carving of the auroral arc by Fridtjof Nansen (from Kaila 1998).
Below: auroral arc at Utsjoki (www.leuku.fi).



Diffuse red aurora: portents of war and bloodshed



Fierce fiery warriors fought upon the
clouds, in ranks and squadrons and
right form of war, that drizzled blood
upon the Capitol.

W. Shakespeare, JC II, 2, 19



Fires in the sky

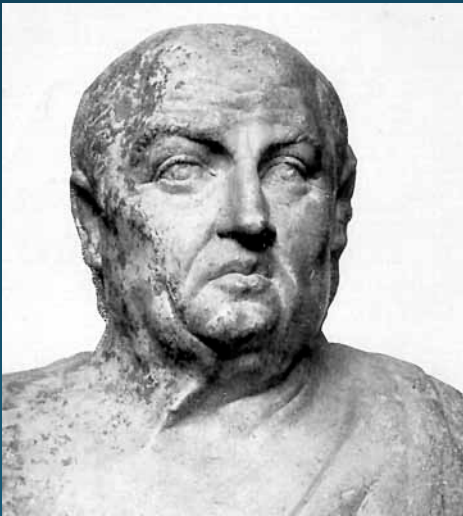


“...the cohorts hurried to the succor of the colony of Ostia, believing it to be on fire. During the greater part of the night the heaven appeared to be illuminated by a faint light.”

Seneca, Naturales Questiones I (37 AD)

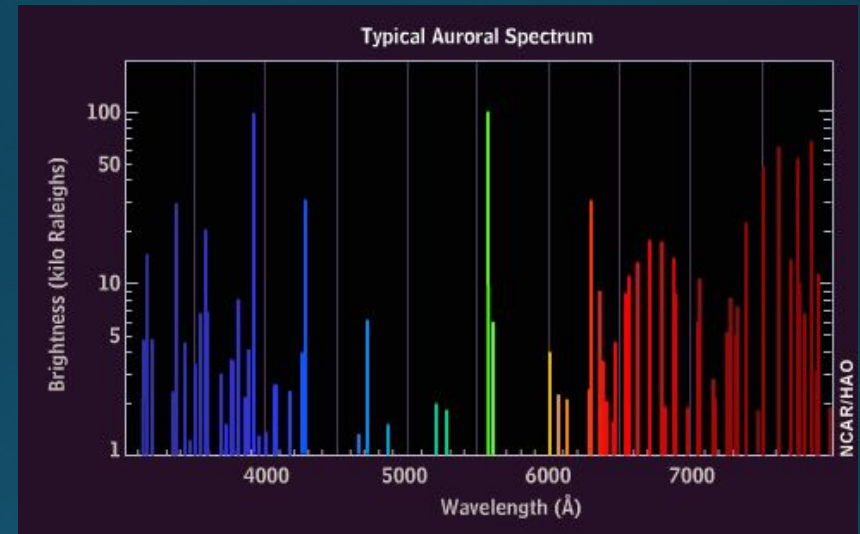
The science of the northern lights

But why does it (the fire) have various colours? Because it makes a difference what element is set ablaze and the quantity and force by which it is set on fire. Falling lights of this sort indicate wind; and, in fact, wind from the region where they started burning. You ask: "The lights which Greeks call sela – how are they produced? In many ways they say. It is possible for the force of the winds to produce them. The high temperature of the upper atmosphere can cause them.



Lucius Annaeus Seneca
Questiones Naturales Book I, Vol. VII

The green line

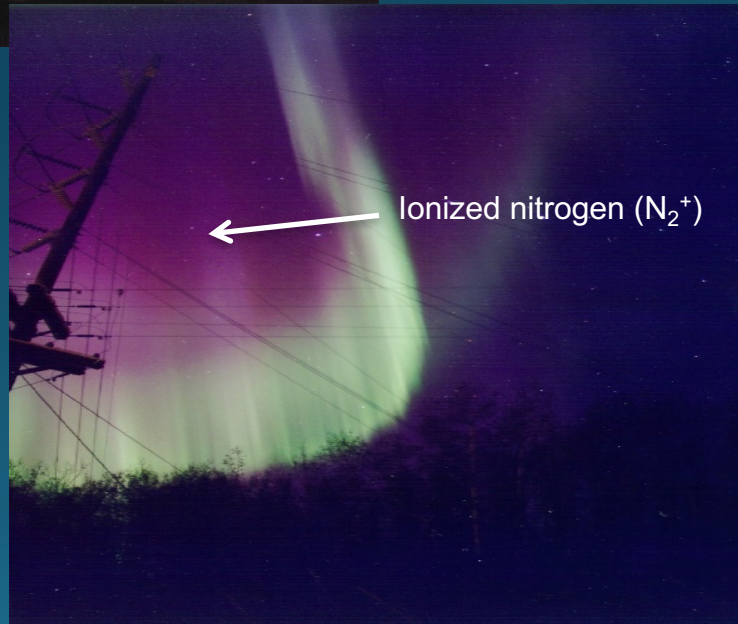
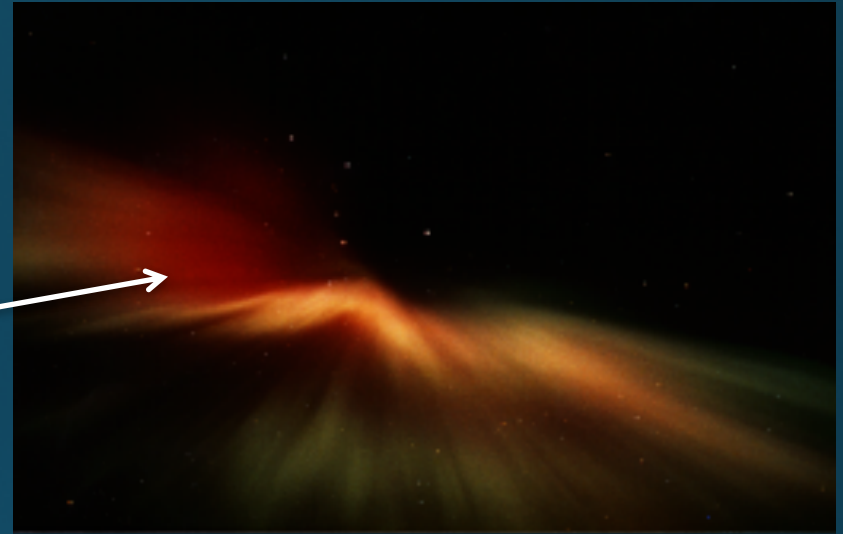
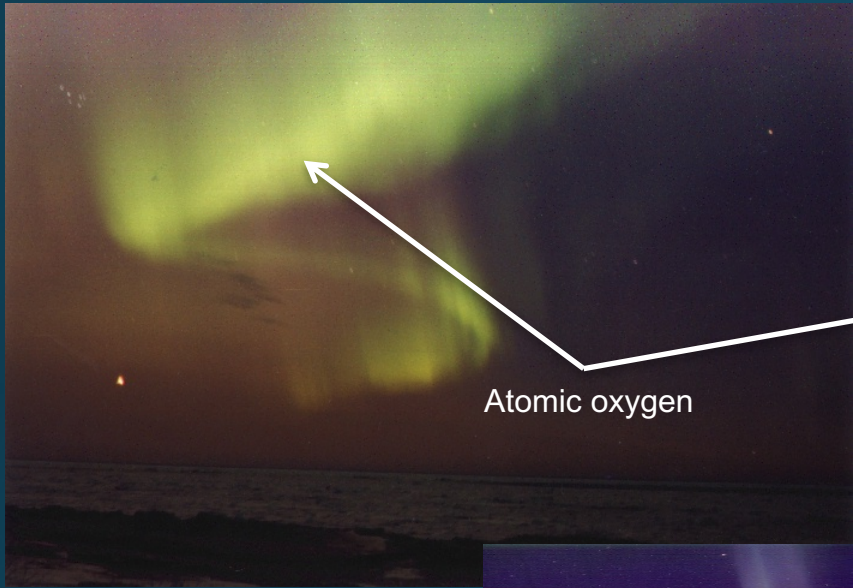


Anders Jonas Ångström (1814–1874) measured the spectrum of the northern lights during 1866 and 1867. In particular he measured the wavelength of 5567 Å for the green line. This was later revised to 5577 Å by Babcock (1923).

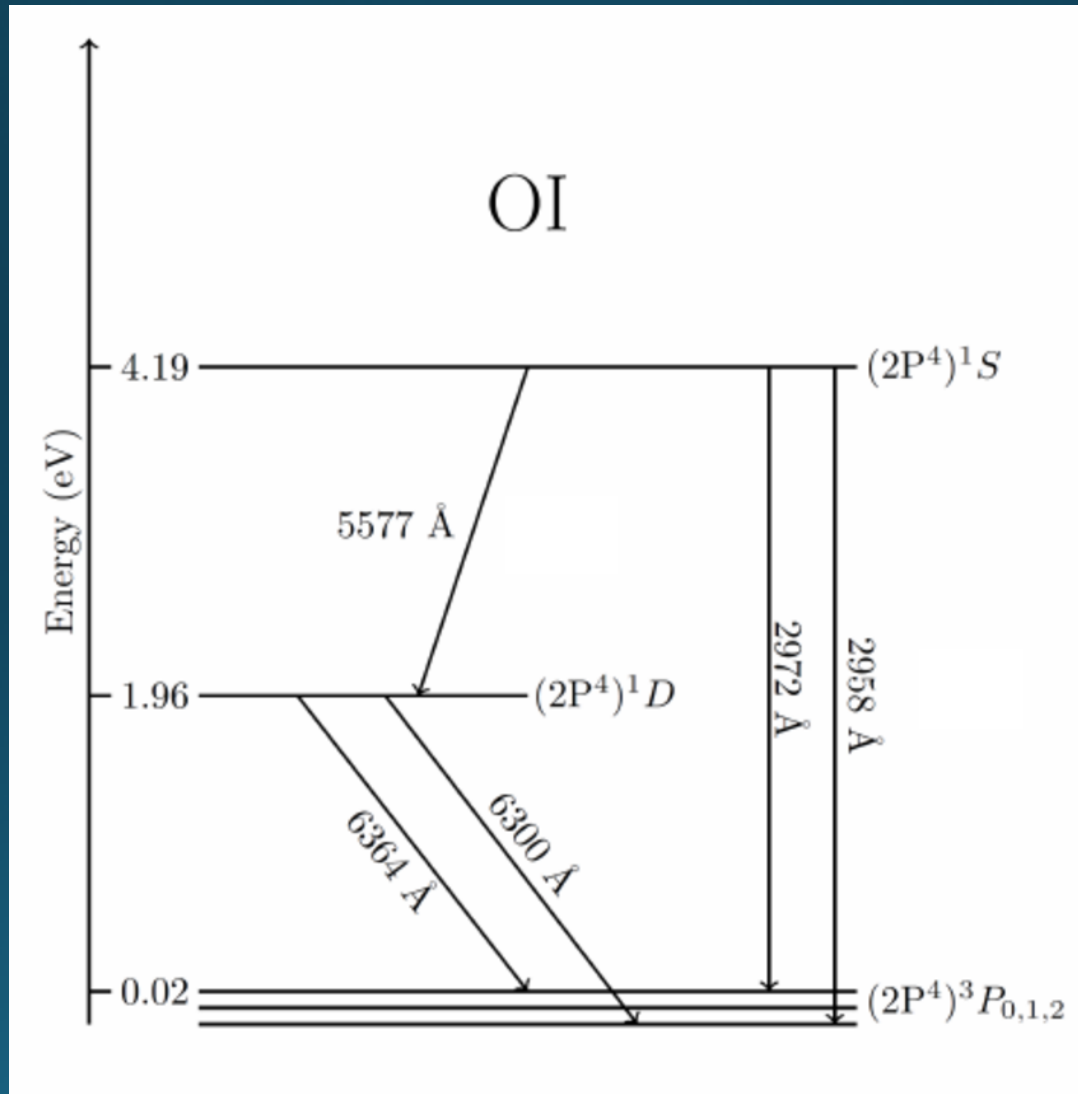
We know nothing at all about the chemical origin of the lines of the polar light – Heinrich Kayser, 1910

- Charles S. Peirce, 1869: unknown species lighter than the hydrogen atom
- Alfred Wegener, 1911: geocoronium
- Niels Bohr, 1913: the Bohr model of atomic hydrogen
- Lars Vegard, 1923: electrified layer of frozen nitrogen (Vegard et al. 1924, *Nature*)
- J. C. McLennan and G. M. Shrum (1924): atomic oxygen
- The samples of Vegard et al. (1924) and McLennan and Shrum (1924) both contaminated by oxygen (Kragh 2009)
- McLennan receives the gold medal from the Royal Society in 1927, his student Shrum is ignored
- Henrikson and Egeland (1988): Theory of the O 5577 Å auroral emissions

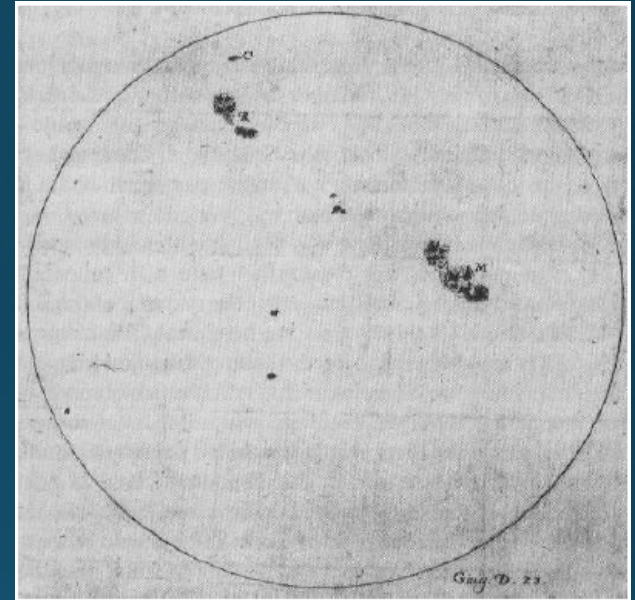
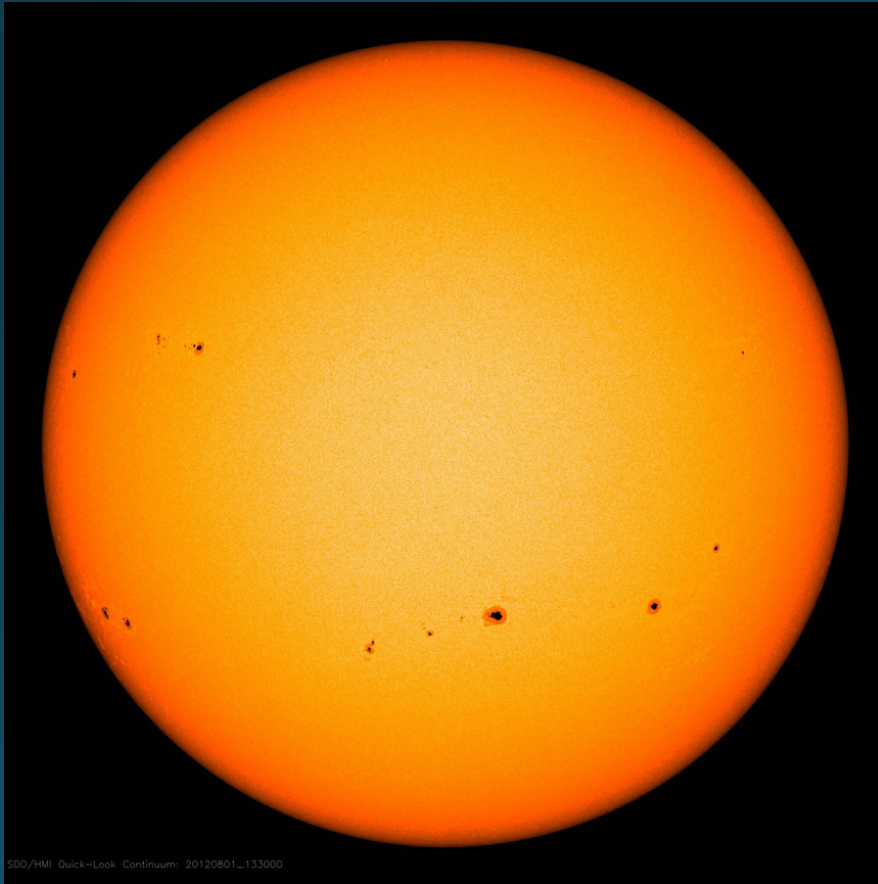
Sources of visible aurora



Oxygen energy level diagram

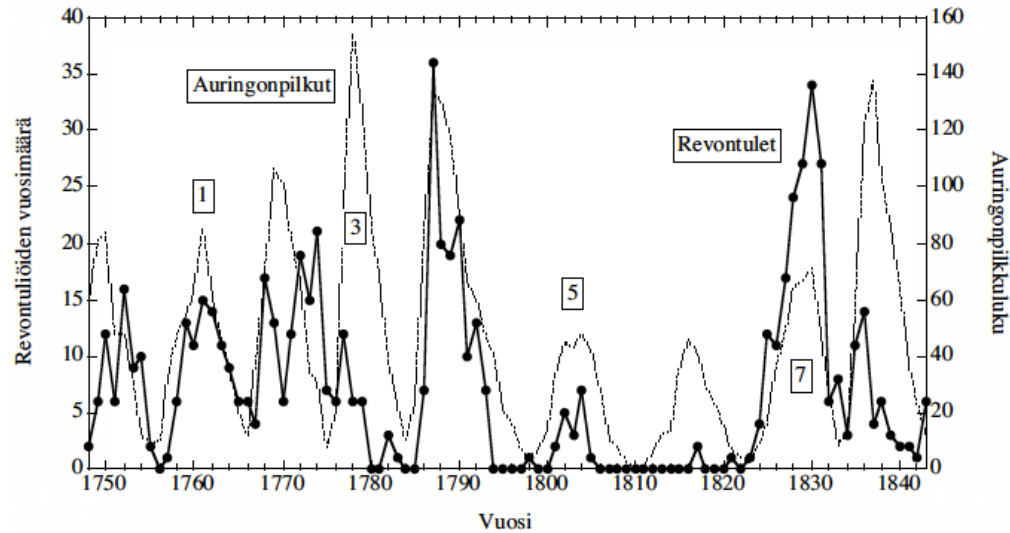


The solar connection

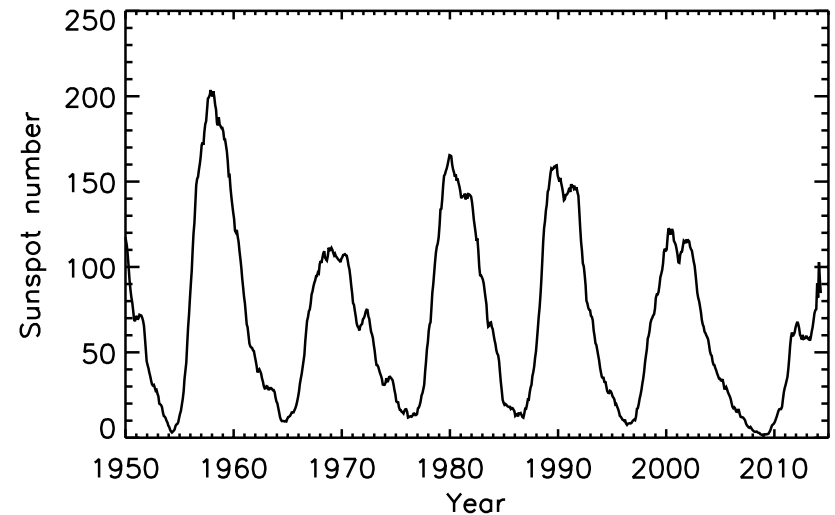


A connection between solar activity and the frequency of the aurora was noted. Above: a sketch of sunspots by Galileo. Left: Image of the sun (July 2012) from NASA's Solar Dynamics Observatory.

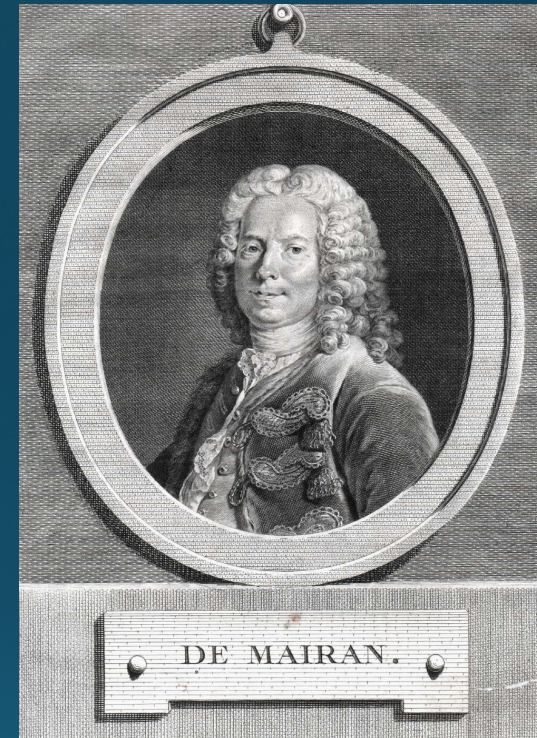
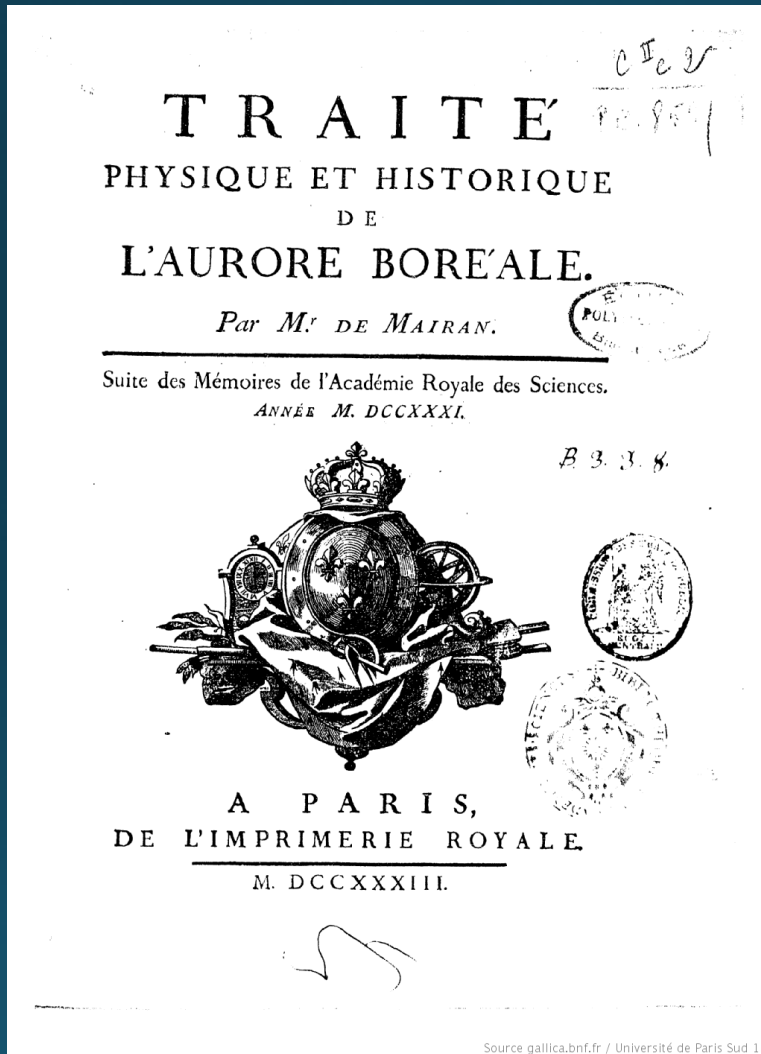
Solar activity and the aurora



Above: Nights with aurora compared with the sunspot number. The observations of the aurora were made by G.G.Hällström in Turku (1748-1828) and Helsinki (1828-1843) (Nevanlinna 2009). Right: The previous sunspot cycle is no. 24.



Falling lights of this sort indicate a wind

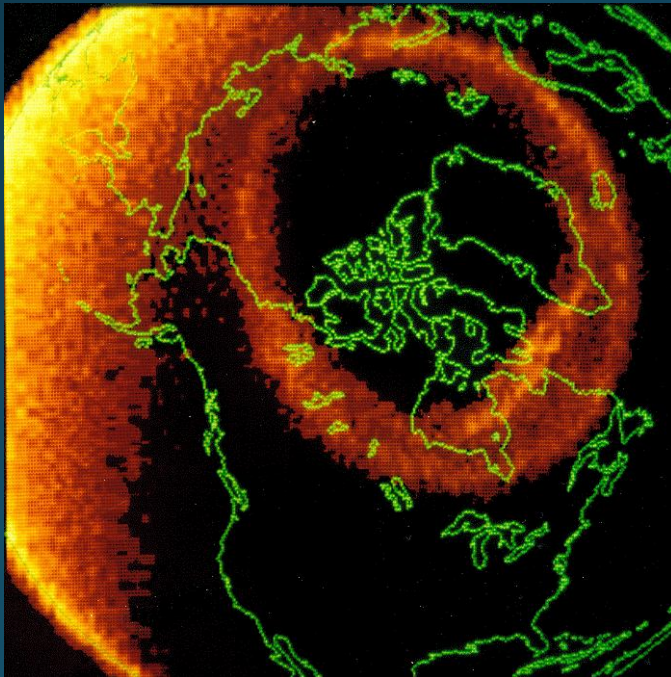
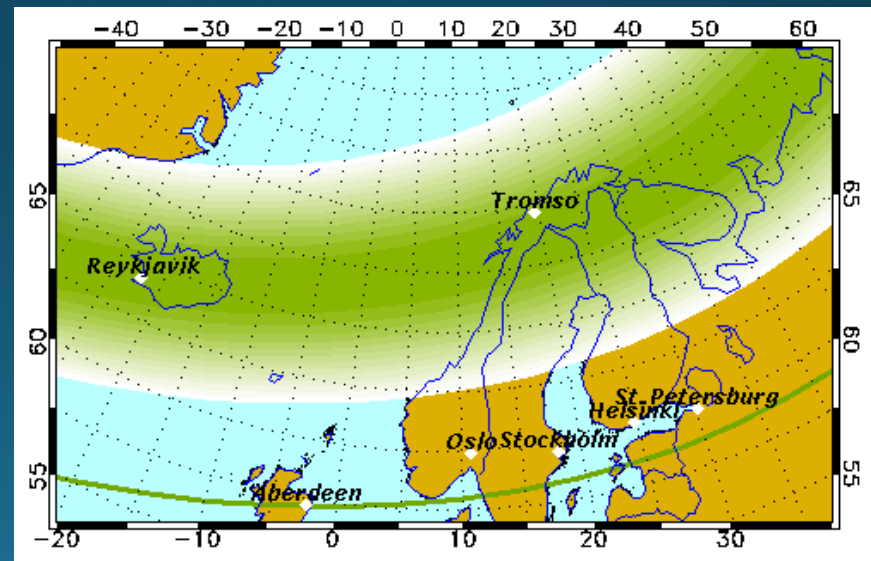
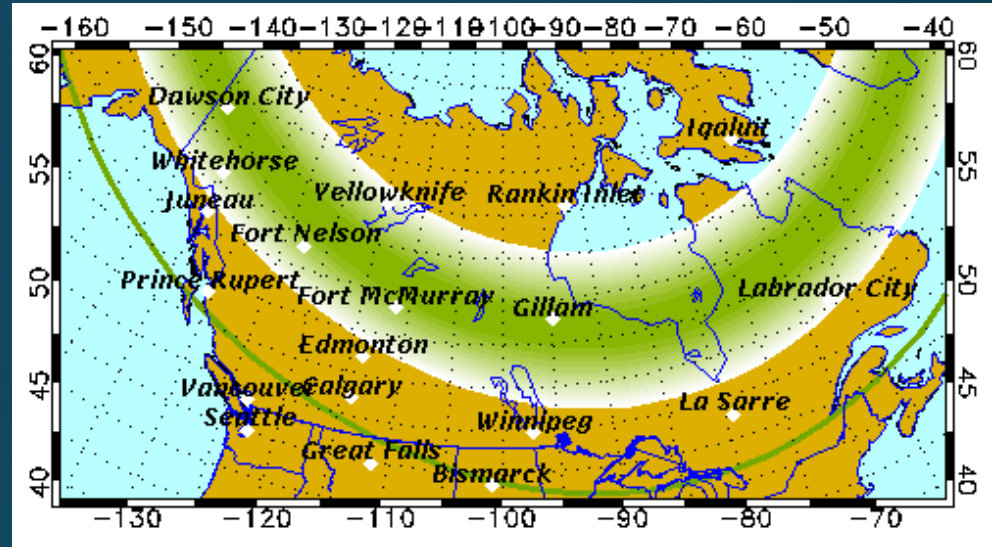


Jean Jacques d'Ortous de Mairan (1678 – 1771) suggested that the interaction of the Earth's atmosphere with the extended atmosphere of the sun produces the aurora borealis.

The auroral oval

In the 19th century aurorae were mapped to a ring surrounding the pole. This ring is not centered at the north pole but the geomagnetic pole.

Image of the UV auroral oval observed by the NASA Dynamics Explorer 1 satellite that was launched in 1981 (below).



The Birkeland – Störmer theory

Kristian Olaf Birkeland (1867 – 1917) argues that the northern lights are caused by electrons as they move along the Earth's magnetic field lines. He also proposes that these electrons stream out from sunspots.

